Supplementary Information

Figure S1. Analytical HPLC coupled to high resolution electrospray ionization mass spectrometer and diode array detector (HPLC/HRESI-MS-DAD) chromatograms for crude samples of dakaramine. The figure shows two peaks (**I** and **II**) with the same UV profile but separated by a column retention time of about 7 minutes. Solvents used for the gradient elution process were A (99.9% H₂O:0.1% HCOOH) and B (99.9% CH₃OH:0.1% HCOOH). The gradient is set to start at 100% A and 0% B at 0 minutes to 0% A and 100% B after 30 minutes at a flow rate of 1.0 mL/min in a Sunfire C18 column (4.6×150 mm, Waters).

Figure S2. HPLC/HRESI-MS-DAD shows peaks I and II to have the same [M+H]⁺.

Figure S3. HPLC/HRESI- MS-DAD data shows peaks I and II with same fragmentation pattern.

Figure S4. Structures/substructures derived for some of the fragments in Figure S3 using ChemDraw Ultra.

Figure S5. Semi-preparative HPLC runs for crude dakaramine fractions shows it is possible for the two peaks **I** and **II** constantly seen in analytical mode (0.1% HCOOH) to separate out or combine to form one peak. Gradients of H₂O:MeOH (100% H₂O to 100% MeOH in 30 minutes and hold for 15 minutes) were used as eluents with flow rate set at 1.5 mL/min on a Sunfire prep C18 column (10 x 250 mm, Waters).

Figure S6. ¹H NMR spectrum of crude fraction containing dakaramine.

Figure S7. ¹³C NMR spectrum of crude fraction containing dakaramine.

Figure S8. ¹H NMR spectrum of pure dakaramine (1).

Figure S9. ¹³C NMR spectrum of pure dakaramine (1).

Figure S10. gHSQCAD NMR spectrum of pure dakaramine (1).

Figure S11. ¹H-¹H gCOSY NMR spectrum of pure dakaramine (1).

Figure S12. HMBCAD NMR spectrum of pure dakaramine (1).

Figure S13. ¹H NMR spectrum of pure 1-hydroxypropan-2-yl acetate (2).

Figure S14. ¹³C NMR spectrum of pure 1-hydroxypropan-2-yl acetate (2).

Figure S15. gHSQCAD NMR spectrum of pure 1-hydroxypropan-2-yl acetate (2).

Figure S16. ¹H-¹H gCOSY NMR spectrum of pure 1-hydroxypropan-2-yl acetate (2).

Figure S17. HMBCAD NMR spectrum of pure 1-hydroxypropan-2-yl acetate (2).

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Figure S2. HPLC/HRESI-MS-DAD shows peaks **I** and **II** to have the same $[M+H]^+$.



Figure S3. HPLC/HRESI-MS-DAD data shows peaks **I** and **II** with same fragmentation pattern.



Figure S4. Structures/sub-structures derived for some of the fragments in Figure S3 using ChemDraw Ultra.

H I

Chemical Formula: C₁₃H₂₀I₂NO⁺ Exact Mass: 459.9629

H

Chemical Formula: C₁₅H₂₄IN₂O⁺ Exact Mass: 375.0928

τH Ĥ

Chemical Formula: C₁₃H₂₁N₂O⁺ Exact Mass: 221.1648



Chemical Formula: C₁₁H₁₆I₂NO⁺ Exact Mass: 431.9316

,Η

Chemical Formula: C₁₂H₁₇INO⁺ Exact Mass: 318.0349



Chemical Formula: C₁₀H₁₂I₂NO⁺ Exact Mass: 415.9003

Chemical Formula: C₁₀H₁₀IO⁺ Exact Mass: 272.9771

Figure S5. Semi-preparative HPLC runs for crude dakaramine fractions shows it is possible for the two peaks **I** and **II** constantly seen in analytical mode (0.1% HCOOH) to separate out or combine to form one peak. Gradients of H₂O:MeOH (100% H₂O to 100% MeOH in 30 minutes and hold for 15 minutes) were used as eluents with flow rate set at 1.5 mL/min on a Sunfire prep C18 column (10 x 250 mm, Waters).





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